

# Nuclear size effects in rotational spectra: a tale with a twist



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with thanks to ....



*Stefan Knecht (Odense)*  
and to Radovan Bast for graphical assistance

# Rovibrational spectroscopy of diatomic molecules

- Within the Born-Oppenheimer (“clamped nuclei”) approximation: **rotating vibrator**

$$\left[ -\frac{\hbar^2}{2\mu} \frac{d}{dR^2} + E^{el}(R) + \frac{\hbar^2 J(J+1)}{2\mu R^2} \right] \psi_{\nu,J}(R) = E_{\nu,J} \psi_{\nu,J}(R) \quad (1)$$

- Solutions were provided by Dunham in 1932 in the framework of the Jeffreys–Wentzel–Kramers–Brillouin (JWKB) approximation<sup>1</sup>

$$E_{\nu J} = h \sum_{kl} Y_{kl} \left( \nu + \frac{1}{2} \right)^k J^l (J+1)^l$$

where  $\nu$  is the vibrational and  $J$  the rotational quantum number.

$$Y_{10} = \frac{\omega_e}{2\pi} \left[ 1 + \left( B_e^2 / 4\omega_e^2 \right) [\dots] \right] \approx \nu_e \quad (\text{harmonic oscillator})$$

$$Y_{01} = B_e \left[ 1 + \left( B_e^2 / 2\omega_e^2 \right) [\dots] \right] \approx B_e \quad (\text{rigid rotator})$$

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<sup>1</sup>J. L. Dunham, Phys. Rev. **41**(1932) 721

# Digression: James Lawrence Dunham



SEPTEMBER 15, 1932

PHYSICAL REVIEW

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## The Energy Levels of a Rotating Vibrator

By J. L. DUNHAM

Harvard University

(Received May 11, 1932)

The energy levels of a rotating vibrator are calculated in considerable detail by means of the Wentzel-Brillouin-Kramers method. The new terms determined are  $\omega_e$  and a set of correction terms which appear in the earlier members of the equation. These correction terms enter in such a way that  $\omega_e$  is not exactly the coefficient of  $(v+\frac{1}{2})$ ;  $B_e$  is not exactly the coefficient of  $K(K+1)$ , etc. However the differences are small and are detectable only in the case of light molecules. The correction terms are of the magnitude of  $B_e^2/\omega_e^2$ . Formulas for the effect of the correction terms on isotope shifts are given, and for the calculation of the correction terms themselves. Also a method is given for obtaining actual potential functions from band spectrum data, based on Morse's potential function. Finally the numerical magnitude of the correction terms for several states of  $H_2$  and for  $NaH$  is discussed.

**1904** Born March 27 in New York; attended Scarsdale High School

**1929** PhD at Harvard University:

*Intensity of lines in the harmonic band of hydrogen chloride in the near infra-red*

**1933** instructor in physics at Harvard;

dies May 3 in Cambridge (Mass.) from pulmonary infarct following appendectomy

## Isotopic mass shift

- Dunham (1932)

$$\left. \begin{array}{lcl} \omega_e & = & \sqrt{\frac{k_e}{\mu}} \\ B_e & = & \frac{\hbar}{2\mu R_e^2} \end{array} \right\} \Rightarrow Y_{kl} = \mu^{-(k/2+l)} U_{kl}$$

- Watson(1980)<sup>2</sup>

$$Y_{kl} = \mu^{-(k/2+l)} U_{kl} \left[ 1 + m_e \left( \frac{\Delta_{kl}^A}{M_A} + \frac{\Delta_{kl}^B}{M_B} \right) \right]$$

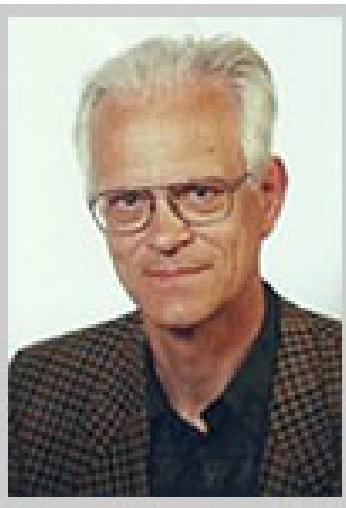
$\Delta_{kl}^X$  - mass-independent atom-specific correction factors

- $\Delta_{kl}^a$ : adiabatic contribution: mass-dependent potential  $E^{el}(R)$
- $\Delta_{kl}^{na}$ : non-adiabatic contribution: coupling of electronic states

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<sup>2</sup>A. H. M. Ross, R. S. Eng, H. Kildal, Opt. Comm. **12** (1971) 433; J. K. G. Watson, J. Mol. Spectr. **80** (1980) 411

# Anomalies



Chemical Physics 67 (1982) 133–138  
North-Holland Publishing Company

## OBSERVED ADIABATIC CORRECTIONS TO THE BORN–OPPENHEIMER APPROXIMATION FOR DIATOMIC MOLECULES WITH TEN VALENCE ELECTRONS

E. TIEMANN

*Institut A für Experimentalphysik, Universität Hannover, D-3000 Hannover, FRG*

and

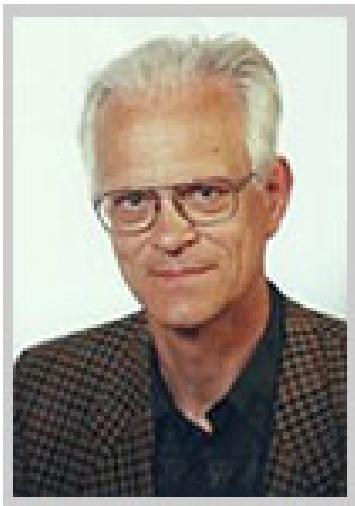
H. ARNST, W.U. STIEDA, T. TÖRRING and J. HOEFT

*Institut für Molekulphysik, Freie Universität Berlin, D-1000 Berlin 33, FRG*

Molecule/atom	$\Delta_{01}$	$\Delta_{01}^{\text{Dunham}}$	$\Delta_{01}^{\text{nonad}}$	$\Delta_{01}^{\text{ad}}$	$r_e^{\text{BO}} (\text{\AA})$
parameters belonging to atoms of the group III and VI					
CO/C	-2.061(34) <sup>a)</sup>	-0.01526(6) <sup>b)</sup>	-1.8484(36) <sup>b)</sup>	-0.197(35)	1.1282291(14)
CS/C	-2.596(49) <sup>c)</sup>	-0.0148	-2.4758 <sup>d)</sup>	-0.105(55)	1.5348224(23)
SiS/Si	-1.392(59) <sup>e)</sup>	-0.0106	-1.1760 <sup>f)</sup>	-0.205(69)	1.9292639(19)
GeS/Ge	-1.463(70) <sup>g)</sup>	+0.0008	-1.2244 <sup>f)</sup>	-0.239(70)	2.0120431(10)
GeSe/Ge	-1.612(46) <sup>g)</sup>	-0.0086	-1.2812 <sup>f)</sup>	-0.322(46)	2.1346018(10)
SnS/Sn	-1.76(19)	+0.0058	-1.0608 <sup>f)</sup>	-0.70(19)	2.2089829(22)
SnSe/Sn	-1.555(84)	-0.0076	-1.1290 <sup>f)</sup>	-0.419(84)	2.3255738(11)
SnTe/Sn	-1.749(97)	-0.0114	-1.1962 <sup>f)</sup>	-0.541(97)	2.5227979(13)
PbS/Pb	-12.94(141) <sup>h)</sup>	+0.0090	-1.2056 <sup>f)</sup>	-11.74(141)	2.2867844(40)
PbSe/Pb	-11.86(92)	-0.0236	-1.1726 <sup>f)</sup>	-10.66(92)	2.4021776(24)
PbTe/Pb	-11.98(81)	-0.0366	-1.1596 <sup>f)</sup>	-10.78(81)	2.5949253(27)
GaF/Ga	-0.60(30)	-0.093	-0.4384 <sup>d)</sup>	-0.07(30)	1.7743351(25)
Gal/Ga	-0.706(96) <sup>i)</sup>	-0.0182	-0.50(30) <sup>k)</sup>	-0.19(30)	2.5746263(40)
InI/In	-2.68(27)	-0.0066	-0.30(30) <sup>k)</sup>	-2.37(45)	2.753618(10)
TlF/Tl	-18.76(110)	+0.0100	-0.1516 <sup>d)</sup>	-18.61(110)	2.084386(10)
TlCl/Tl	-18.96(200)	-0.0040	-0.1860 <sup>d)</sup>	-18.76(200)	2.484739(10)
TlBr/Tl	-15.61(46)	-0.0216	-0.05(30) <sup>k)</sup>	-15.55(75)	2.6181114(15)
TlI/Tl	-14.68(47)	-0.0172	-0.05(30) <sup>k)</sup>	-14.61(75)	2.813614(10)

Unusually large adiabatic mass corrections observed for lead/thallium compounds

# Isotopic field shift



Chemical Physics 68 (1982) 21–28  
North-Holland Publishing Company

## ISOTOPIC FIELD SHIFT OF THE ROTATIONAL ENERGY OF THE Pb-CHALCOGENIDES AND Ti-HALIDES

J. SCHLEMBACH and E. TIEMANN

*Institut A für Experimentalphysik, Universität Hannover, D-3000 Hannover, FRG*

$$Y_{kl} = \mu^{-(k/2+l)} U_{kl} \left[ 1 + m_e \left( \frac{\Delta_{kl}^A}{M_A} + \frac{\Delta_{kl}^B}{M_B} \right) + V_{kl}^A \delta \langle r^2 \rangle_{AA'} + V_{kl}^B \delta \langle r^2 \rangle_{BB'} \right]$$

$\delta \langle r^2 \rangle_{XX'}$  change in mean squared nuclear charge radius between isotope X' and reference isotope X

$V_{kl}^X$  correction factor due to the finite size of nucleus X

## Isotopic field shift on rotational constant

Focus on the rotational constant

$$Y_{01} = \mu^{-1} U_{01} \left[ 1 + m_e \left( \frac{\Delta_{01}^A}{M_A} + \frac{\Delta_{01}^B}{M_B} \right) + V_{01}^A \delta \langle r^2 \rangle_{AA'} + V_{01}^B \delta \langle r^2 \rangle_{BB'} \right]$$

for which the correction factor reads

$$V_{01}^X = \frac{Z_X e^2}{3\epsilon_0 k_e R_e} \left( \frac{d\rho_0^X}{dR} \right)_{R_e} ; \quad X = A, B$$

Table 2

Differential coefficient of the electron density from the function of the internuclear distance at the nucleus Pb or Tl, calculated with the values of table 1

Molecule AB	$(d\rho/dR)_{R_e}^A (\text{\AA}^{-4})$	Molecule AB	$(d\rho/dR)_{R_e}^A (\text{\AA}^{-4})$
PbS	213(17)	TlF	254(30)
PbSe	177(15)	TlCl	185(25)
PbTe	148(11)	TlBr	142.0(42)
		TlI	120.6(38)

# A computational study

S. A. Cooke, M. C. L. Gerry and D. P. Chong, Chem. Phys. **298** (2004) 205:

Density functional theory (DFT) calculations using the long-range corrected functional SAOP and the Zeroth Order Regular Approximation (ZORA) Hamiltonian



Derivative of the electron density at the metal nucleus with respect to the internuclear separation calculated using the SAOP method at the equilibrium bond length in units of  $\text{\AA}^{-4}$

Molecule	SAOP				Experiment [12,13] <sup>a</sup>
	DZ	TZP	TZ2P	QZ4P	
PbO				40.3	_ <sup>b</sup>
PbS	40.8	31.7	67.3	27.8	23.19(46)
PbSe	38.8	20.1	25.5	25.6	17.7(15)
PbTe	39.5	23.7	31.5	21.1	14.8(11)
TlF	71.8	97.1	107.3	44.4	25.4(30)
TlCl	35.6	45.8	73.8	36.7	18.5(25)
TlBr	28.8	32.4	46.8	26.5	14.20(42)
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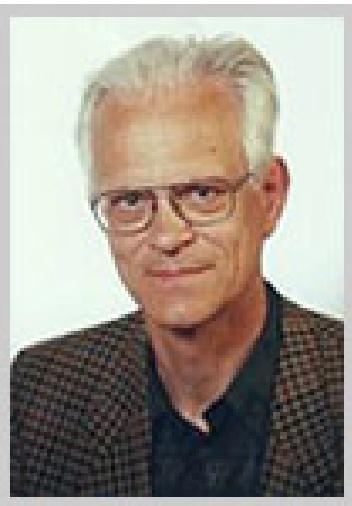


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<sup>a</sup> Experimental values taken from [12] have been corrected in light of an error pointed out in [13].

# A mysterious correction factor



Chemical Physics 93 (1985) 349–358  
North-Holland, Amsterdam

## MOLECULAR-BEAM-LASER STUDIES OF THE STATES $X^1\Sigma^+$ AND $A0^+$ OF PbS

H. KNÖCKEL, T. KRÖCKERTSKOTHEN and E. TIEMANN

*Institut für Atom- und Molekulphysik, Universität Hannover, D-3000 Hannover, FRG*

- 
- ★ There is a computational error in table 2 of ref. [2]: the value of  $\rho'_{\text{cl}}$  should be reduced by a factor of 10.

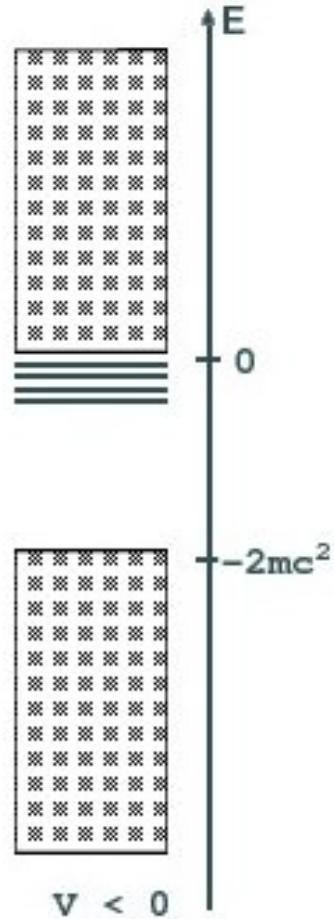
We note that the expression for the finite nuclear size correction factor is modified:

$$V_{01}^X \rightarrow \pi^2 V_{01}^X$$

# 4-component relativistic Hamiltonians

- Generic form of electronic Hamiltonian:

$$H = \sum_i h(i) + \frac{1}{2} \sum_{i \neq j} g(i, j) + V_{NN}; \quad h(i) = h_0 + V_{eN}$$



- One-electron operator: Dirac operator in the molecular field

$$\hat{h}_D(i) = \beta'_i mc^2 + c(\boldsymbol{\alpha}_i \cdot \mathbf{p}_i) + \hat{V}_{eN}; \quad \beta' = \beta - 1$$

- Two-electron operator: (Coulomb gauge)

$$\begin{aligned}
 \hat{g}(i, j) &= \frac{1}{r_{ij}} \\
 &- \frac{c\boldsymbol{\alpha}_i \cdot c\boldsymbol{\alpha}_j}{c^2 r_{ij}} - \frac{(c\boldsymbol{\alpha}_i \cdot \nabla_i)(c\boldsymbol{\alpha}_j \cdot \nabla_j) r_{ij}}{2c^2} \\
 &+ \dots
 \end{aligned}$$

## From 4 to 2 components

- Unitary transformation: decoupling and renormalization

$$\hat{U}^\dagger \begin{bmatrix} V & c(\boldsymbol{\sigma} \cdot \mathbf{p}) \\ c(\boldsymbol{\sigma} \cdot \mathbf{p}) & V - 2mc^2 \end{bmatrix} \hat{U} = \begin{bmatrix} \hat{h}_+ & 0 \\ 0 & \hat{h}_- \end{bmatrix}; \quad \hat{U} = \begin{bmatrix} \Omega_+ & -\chi^\dagger \Omega_- \\ \chi \Omega_+ & \Omega_- \end{bmatrix}$$

- Transformation of orbitals:

$$\begin{bmatrix} \psi^L \\ \psi^S \end{bmatrix} \rightarrow \begin{bmatrix} \Omega_+ [\psi^L + \chi^\dagger \psi^S] \\ \Omega_- [\psi^S - \chi \psi^L] \end{bmatrix}$$

- Positive-energy solutions:  $\chi$  is the exact coupling of the large and small components

$$\psi^S = \chi \psi^L; \quad \chi = (2mc^2 - V + E)^{-1} c(\boldsymbol{\sigma} \cdot \mathbf{p})$$

- Approximate Hamiltonians, e.g. ZORA

$$\chi = \frac{c}{2mc^2 - V} \left[ 1 + \frac{E^+}{2mc^2 - V} \right]^{-1} (\boldsymbol{\sigma} \cdot \mathbf{p}) \sim \frac{c}{2mc^2 - V} (\boldsymbol{\sigma} \cdot \mathbf{p})$$

# Decoupling the easy way

M. Iliaš and T. Saue , J. Chem. Phys. **126** (2007) 064102;

W. Kutzelnigg and W. Liu, *J. Chem. Phys.* **123** (2005) 241102;

M. Iliaš, H. J. Aa. Jensen, V. Kellö, B. O. Roos and M. Urban, *Chem. Phys. Lett.* **408** (2005) 210

1. Solve the Dirac equation on matrix form
2. Extract the coupling  $\chi$  from the solutions
3. Construct the transformation matrix  $U$ , next  $h^{X2C}$

## Advantages:

- reproduces exactly the positive-energy spectrum of the Dirac Hamiltonian
- all matrix manipulations; no new operators to program
- explicit representation of transformation matrix;  
any property operator can be transformed on the fly, no picture change errors

**Note:** SO-corrections to the 2-electron interaction are handled by the AMFI approach  
(B. A. Heß, C. M. Marian, U. Wahlgren and O. Gropen, *Chem. Phys. Lett.* **251** (1996) 365)

## Picture change errors



The 2-component Hamiltonian is obtained as

$$H^{2c} = [U^\dagger H^{4c} U]_{++}$$

Property operators  $\Omega^{4c}$  must be subjected to the same decoupling transformation as the Hamiltonian, that is

$$\Omega^{2c} = [U^\dagger \Omega^{4c} U]_{++}$$

Use of the approximate expression

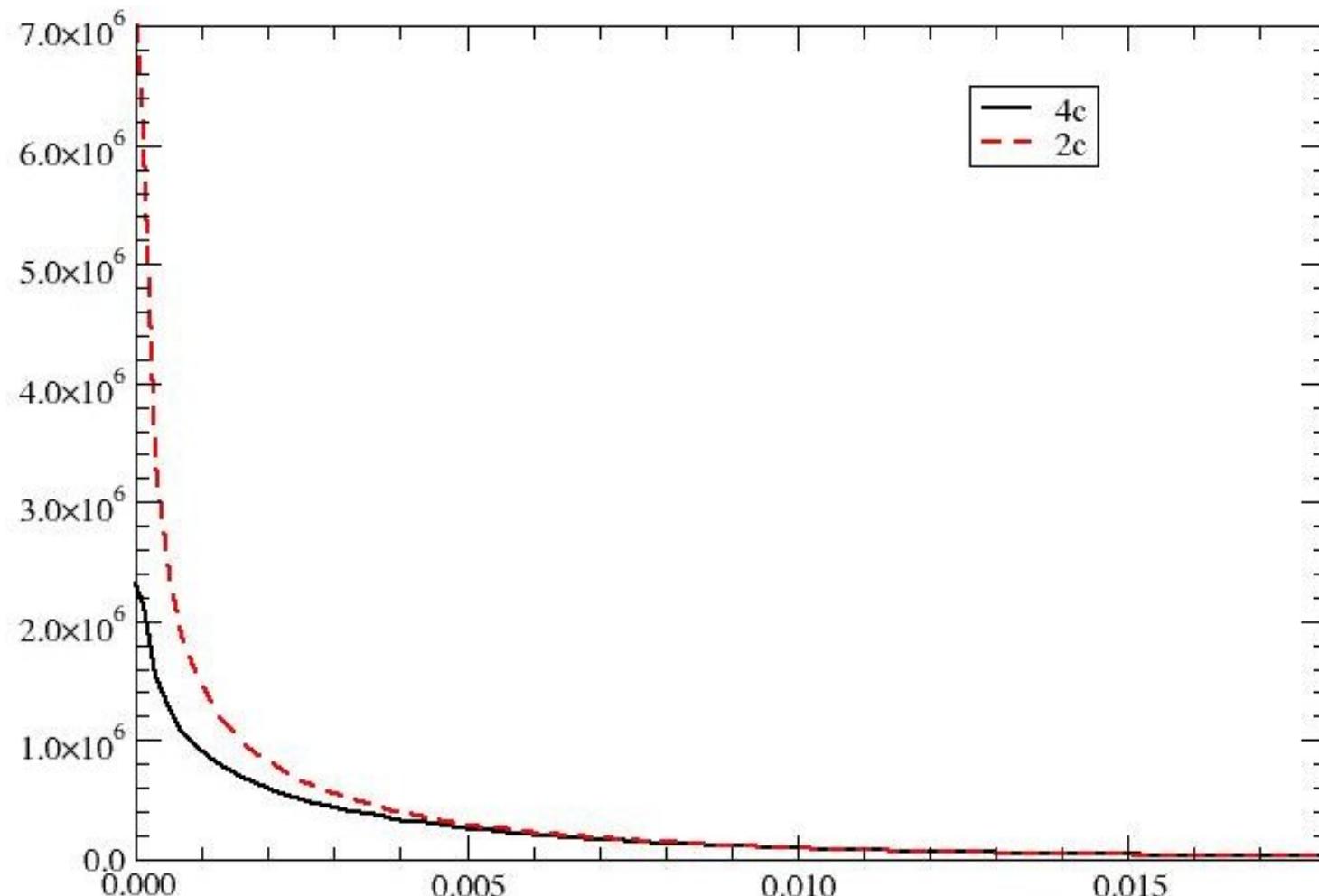
$$\Omega^{2c} \approx [\Omega^{4c}]_{LL}$$

leads to **picture change errors** that may be larger than the relativistic effects !

## An example: the electron density

$$\rho^{4c}(\mathbf{P}) = -e \sum_i \langle \psi_i^{4c} | \delta(\mathbf{r} - \mathbf{P}) | \psi_i^{4c} \rangle = -e \sum_i \psi_i^{4c\dagger}(\mathbf{P}) \psi_i^{4c}(\mathbf{P})$$

$$\rho^{2c}(\mathbf{P}) = -e \sum_i \left\langle \psi_i^{2c} \left| [U^\dagger \delta(\mathbf{r} - \mathbf{P}) U]_{++} \right| \psi_i^{2c} \right\rangle \neq -e \sum_i \psi_i^{2c\dagger}(\mathbf{P}) \psi_i^{2c}(\mathbf{P})$$

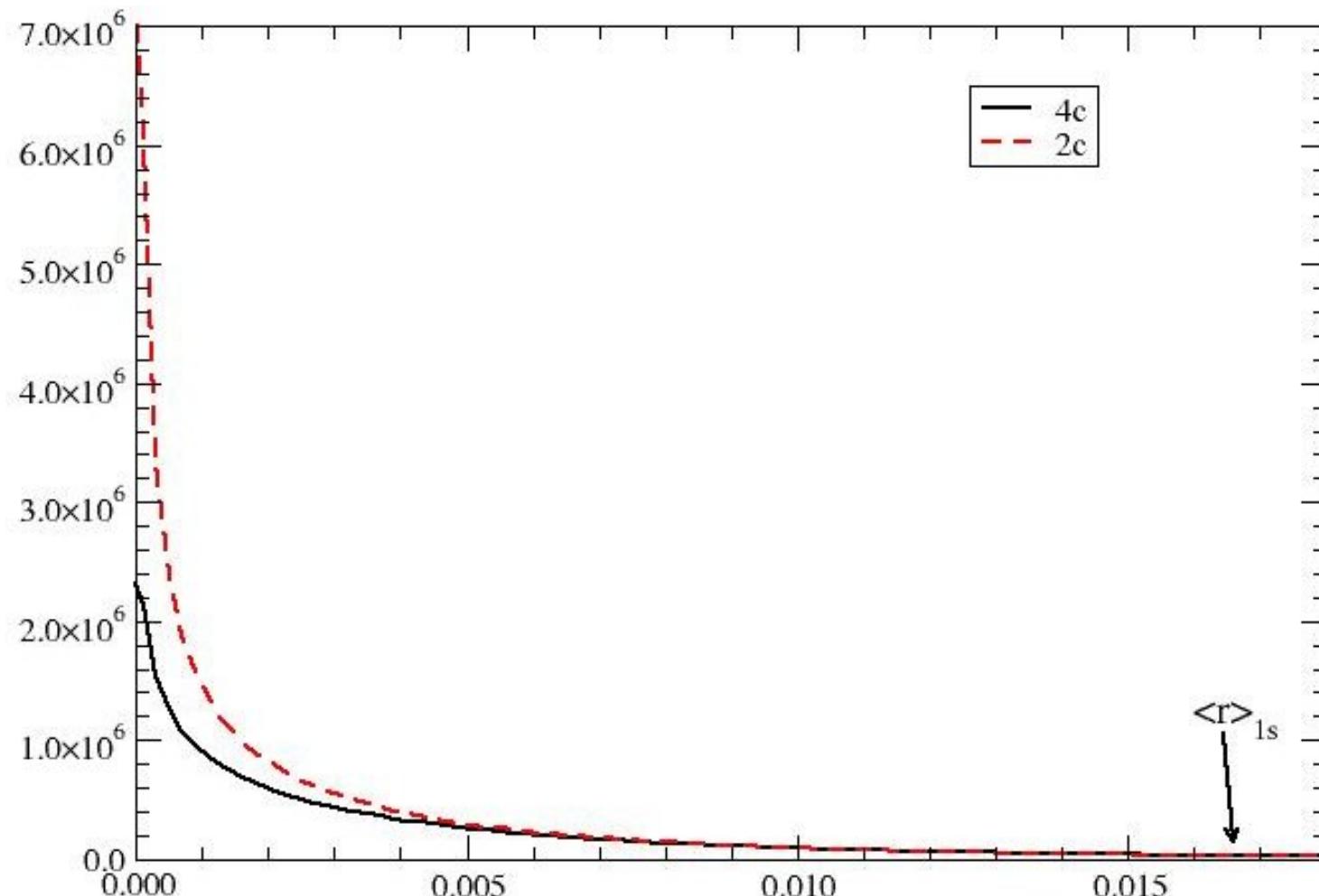


$\sum_i \psi_i^{4c\dagger} \psi_i^{4c}$  vs.  $\sum_i \psi_i^{2c\dagger} \psi_i^{2c}$  for the mercury atom

## An example: the electron density

$$\rho^{4c}(\mathbf{P}) = -e \sum_i \langle \psi_i^{4c} | \delta(\mathbf{r} - \mathbf{P}) | \psi_i^{4c} \rangle = -e \sum_i \psi_i^{4c\dagger}(\mathbf{P}) \psi_i^{4c}(\mathbf{P})$$

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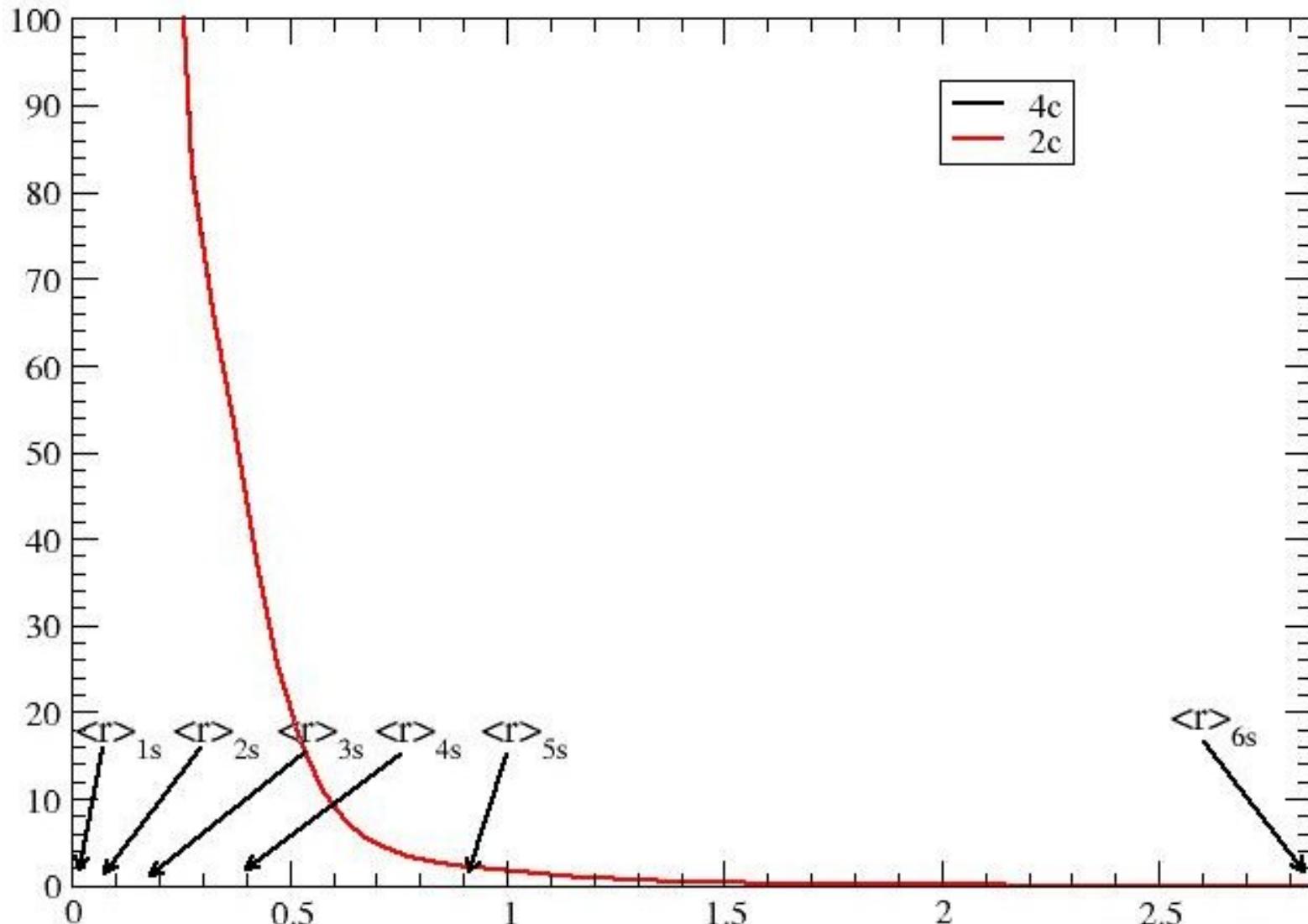


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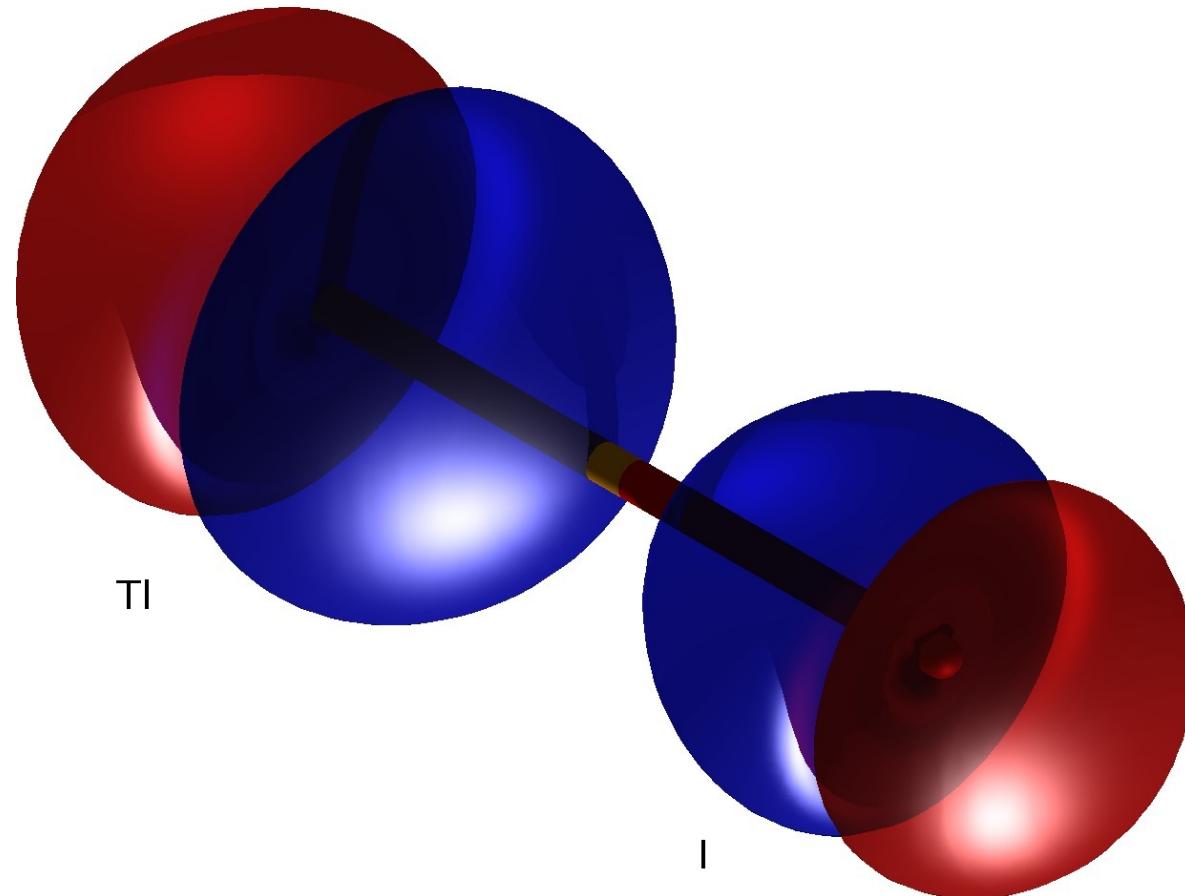
$$\rho^{2c}(\mathbf{P}) = -e \sum_i \left\langle \psi_i^{2c} \left| [U^\dagger \delta(\mathbf{r} - \mathbf{P}) U]_{++} \right| \psi_i^{2c} \right\rangle \neq -e \sum_i \psi_i^{2c\dagger}(\mathbf{P}) \psi_i^{2c}(\mathbf{P})$$



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# Thallium iodine ( $R_e = 2.8136 \text{ \AA}$ ): relativistic effects

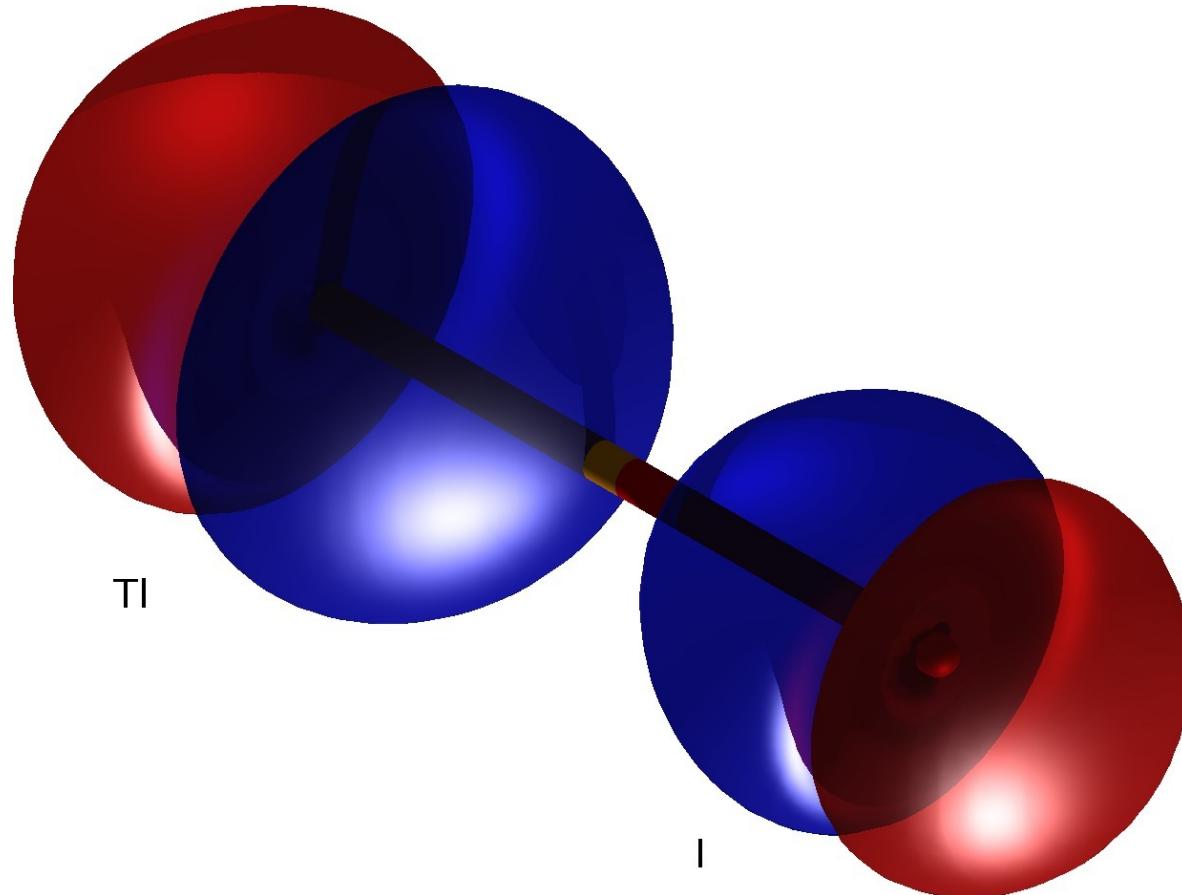
Method	Basis	Hamiltonian	Approach	$\rho_0^{Tl}(\text{\AA}^{-3})$	$d\rho_0^{Tl}/dR _{R=R_e}(\text{\AA}^{-4})$
SAOP	TZ	4c	$\langle \rho^{Tl} \rangle$	17153354.69	118.14
SAOP	TZ	NR	$\langle \rho^{Tl} \rangle$	2535515.16	24.50
				-85.2%	-79.3%



## Thallium iodine ( $R_e = 2.8136 \text{ \AA}$ ): picture change errors

Method	Basis	Hamiltonian	Approach	$\rho_0^{Tl}(\text{\AA}^{-3})$	$d\rho_0^{Tl}/dR _{R=R_e}(\text{\AA}^{-4})$
SAOP	TZ	4c	$\langle \rho^{Tl} \rangle$	17153354.69	118.14
SAOP	TZ	X2C	$\psi^\dagger \psi$	54950057.82	788.15
				+220.3%	+567.1%

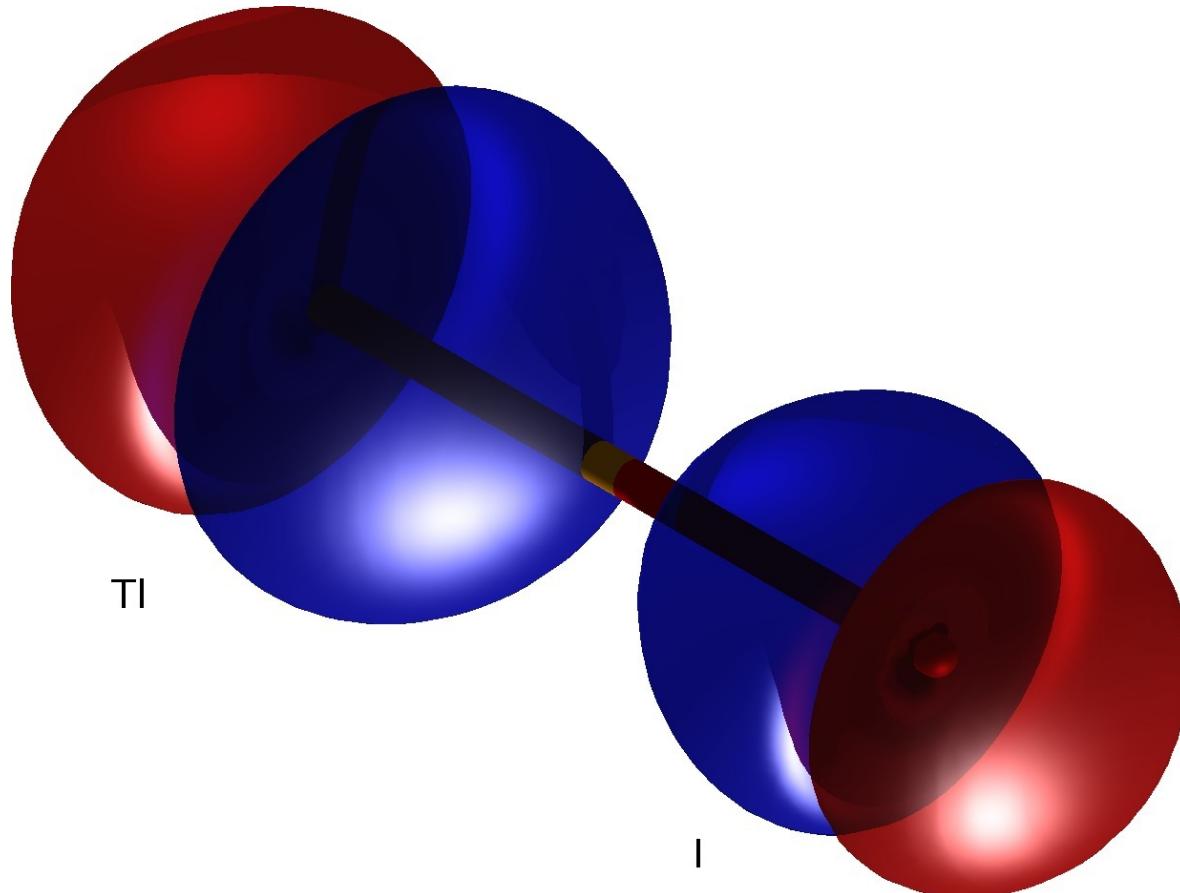
The picture change error is larger than the relativistic effect !



## Thallium iodine ( $R_e = 2.8136 \text{ \AA}$ ): picture change errors

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SAOP	TZ	X2C	$\langle \rho^{Tl} \rangle$	17153358.65	118.24

The picture change error is eliminated by correct transformation of the property operator.



## Thallium iodine ( $R_e = 2.8136 \text{ \AA}$ ): comparing with experiment

Method	Basis	Hamiltonian	Approach	$\rho_0^{Tl}(\text{\AA}^{-3})$	$d\rho_0^{Tl}/dR _{R=R_e}(\text{\AA}^{-4})$
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Table 2

Differential coefficient of the electron density from the function of the internuclear distance at the nucleus Pb or Tl, calculated with the values of table 1

Molecule AB	$(d\rho/dR)_{R_e}^A(\text{\AA}^{-4})$	Molecule AB	$(d\rho/dR)_{R_e}^A(\text{\AA}^{-4})$
PbS	213(17)	TlF	254(30)
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# Thallium iodine ( $R_e = 2.8136 \text{ \AA}$ ): comparing with calculations

Method	Basis	Hamiltonian	Approach	$\rho_0^{Tl}(\text{\AA}^{-3})$	$d\rho_0^{Tl}/dR _{R=R_e}(\text{\AA}^{-4})$
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<sup>a</sup> Experimental values taken from [12] have been corrected in light of an error pointed out in [13].

Cooke *et al.* did non-relativistic calculations....

# Thallium iodine ( $R_e = 2.8136 \text{ \AA}$ ): Benchmark calculations

Method	Basis	Hamiltonian	Approach	$\rho_0^{Tl}(\text{\AA}^{-3})$	$d\rho_0^{Tl}/dR _{R=R_e}(\text{\AA}^{-4})$
HF	TZ2s1p	4C	$\langle \rho^{Tl} \rangle$	17347159.65	212.72
SAOP	TZ2s1p	4C	$\langle \rho^{Tl} \rangle$	17426852.47	120.26
CCSD(T)	TZ2s1p	4C	$\langle \rho^{Tl} \rangle$	17347293.64	142.43

Finite field CCSD(T) calculations. 30 electrons correlated ( Tl: 5d6s6p + I:4d5s5p). Virtual cutoff: 36 a.u.

Only LL integrals included in the valence correlation  
 (with all integrals  $d\rho^{Tl}/dR|_{R=R_e}$  changes by at most  $1 \text{ \AA}^{-4}$ ).

Table 2

Differential coefficient of the electron density from the function of the internuclear distance at the nucleus Pb or Tl, calculated with the values of table 1

Molecule AB	$(d\rho/dR)_{R_e}^A (\text{\AA}^{-4})$	Molecule AB	$(d\rho/dR)_{R_e}^A (\text{\AA}^{-4})$
PbS	213(17)	TlF	254(30)
PbSe	177(15)	TlCl	185(25)
PbTe	148(11)	TlBr	142.0(42)
		TlI	120.6(38)

Note that CCSD(T) takes us away from the experimental result !

## Nuclear volume correction to the rotational constant

The first-order change in the rotational constant due to a change of bond length is

$$B_e = \frac{\hbar}{2\mu R_e^2} \Rightarrow \delta B_e = -\frac{\hbar}{\mu R_e^3} \delta R_e = B_e \left( -\frac{2}{R_e} \right) \delta R_e$$

To determine the change in bond length due to a change of nuclear size of atom A, we consider the electronic energy as a function of internuclear distance  $R$  and nuclear radius  $\xi$  of atom A

$$E^{el} \equiv E^{el}(R, \xi)$$

The equilibrium bond length for given nuclear radius  $\xi$  is found by minimizing the energy with respect to internuclear distance

$$\frac{dE^{el}}{dR} \Big|_{\xi} = 0$$

The equilibrium distance accordingly becomes a function of internuclear distance  $R_e \equiv R_e(\xi)$ , and  $\delta R_e$  can be found by **variational perturbation theory**.

## Nuclear volume correction to the equilibrium bond distance

$$\frac{dE^{el}}{dR} \Big|_{\xi} = 0 \quad \Rightarrow \quad \frac{d^2E^{el}}{d\xi dR} \Big|_{\xi_A} = \left[ \frac{d^2E^{el}}{dR^2} \frac{dR}{d\xi} + \frac{\partial}{\partial \xi} \left( \frac{dE^{el}}{dR} \right) \right]_{\xi_A} = 0$$

---

$$\left[ \frac{d^2E^{el}}{dR^2} \right]_{\xi_A} = k_e^A \quad \text{force constant for reference isotope A}$$

---

The electrostatic interaction of the electrons with nucleus A  
depends on nuclear radius  $\xi$  of atom A

$$\begin{aligned} \frac{\partial}{\partial \xi} \left( \frac{dE^{el}}{dR} \right) &= \frac{\partial}{\partial \xi} \frac{d}{dR} \left[ \int \rho_e(\mathbf{r}_e; R, \xi) \phi_n^A(\mathbf{r}_e; \xi) d^3 r_e \right] \\ &\approx \int \frac{d\rho_e(\mathbf{r}_e; R)}{dR} \frac{\partial \phi_n^A(\mathbf{r}_e; \xi)}{d\xi} d^3 r_e \\ &\approx \frac{d\bar{\rho}_e}{dR} \int \frac{\partial \phi_n^A(\mathbf{r}_e; \xi)}{d\xi} d^3 r_e \end{aligned}$$

(we ignore any modification of  $\rho_e$  due to a change of nuclear size)

# Nuclear models

In the relativistic domain a finite nucleus model is mandatory due to the singularity of the electronic wave function at a point nucleus

**Homogeneously charged sphere**

$$\rho_n^H(\mathbf{r}) = \rho_0^H \Theta(\zeta - r)$$

**Gaussian charge distribution**

$$\rho_n^G(\mathbf{r}) = \rho_0^G \exp[-r^2/\xi_G^2]$$

$$\frac{\partial \phi_n^H}{\partial \xi} = \begin{cases} -\frac{3Z}{2\xi^2} \frac{1}{4\pi\varepsilon_0} \left(1 - \frac{r^2}{\xi^2}\right) & ; \quad r \leq \xi \\ 0 & ; \quad r > \xi \end{cases} \quad \frac{\partial \phi_n^G}{\partial \xi} = -\frac{Z}{2\varepsilon_0} \xi_G \left(\frac{\partial \xi_G}{\partial \xi}\right) \rho_n^G$$

Connecting radii:

$$\langle r_n^2 \rangle = \frac{3}{5} \xi^2 = \frac{3}{2} \xi_G^2 \quad \Rightarrow \quad \xi_G = \xi \sqrt{\frac{2}{5}}$$

Both models then give the same result:

$$\left[ \frac{\partial}{\partial \xi} \left( \frac{dE^{el}}{dR} \right) \right]_{\xi_A} \approx \frac{d\bar{\rho}_e}{dR} \int \frac{\partial \phi_n^A(\mathbf{r}_e; \xi)}{\partial \xi} \Big|_{\xi_A} d^3 r_e = -\frac{4\pi}{5} Z_A \xi_A \frac{d\bar{\rho}_e}{dR}$$

## Nuclear volume correction to the rotational constant

Change in equilibrium distance due to isotope substitution  $A \rightarrow A'$

$$\delta R_e = -\frac{Z_A e^2}{6\varepsilon_0 k_e^A} \delta \langle r_n^2 \rangle_{AA'} \left( \frac{d\bar{\rho}_e}{dR} \right)_{R_e^A}$$

Using literature data we find for TII

$$\delta R_e^{Tl(203,205)} = -6.4 \cdot 10^{-17} m !$$

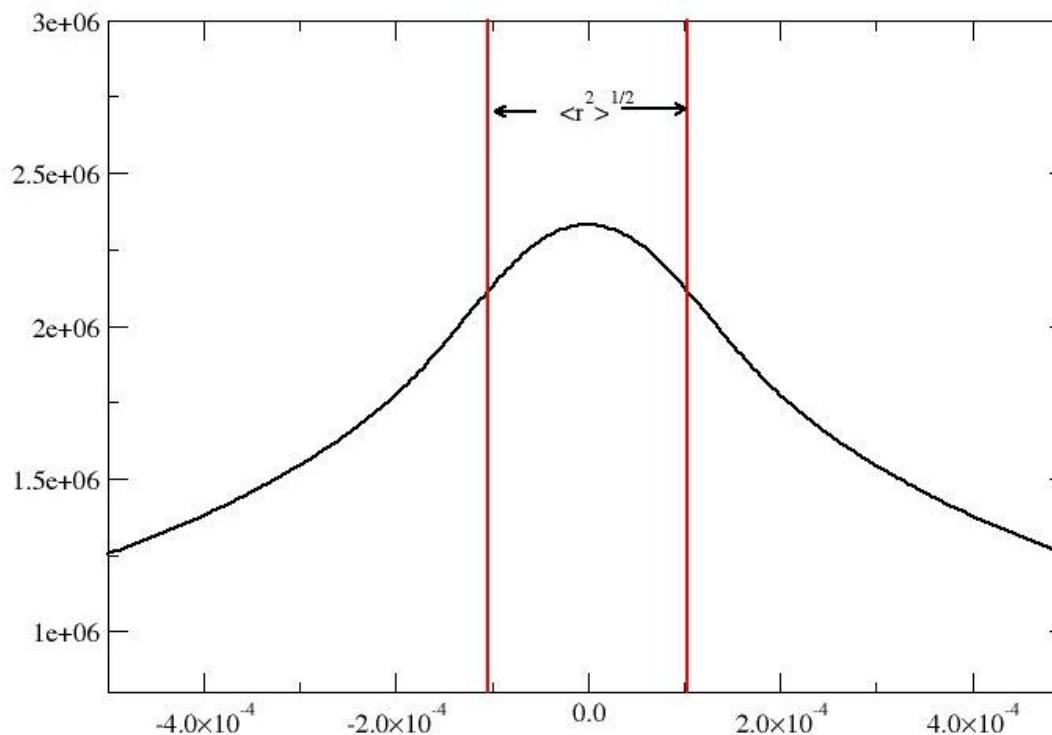
Corresponding change in rotational constant

$$\delta B_e = B_e^A \left( -\frac{2}{R_e} \right) \delta R_e = B_e^A V_{01}^A \delta \langle r_n^2 \rangle_{AA'} ; \quad V_{01}^A = \frac{Z_A e^2}{3\varepsilon_0 k_e^A R_e^A} \left( \frac{d\bar{\rho}_e}{dR} \right)_{R_e^A}$$

## From effective to contact density

Since the effective density is sampled inside the nuclear volume it is typically approximated by the contact density, that is, value of the electronic density at the nucleus

$$\bar{\rho} \approx \rho(0)$$

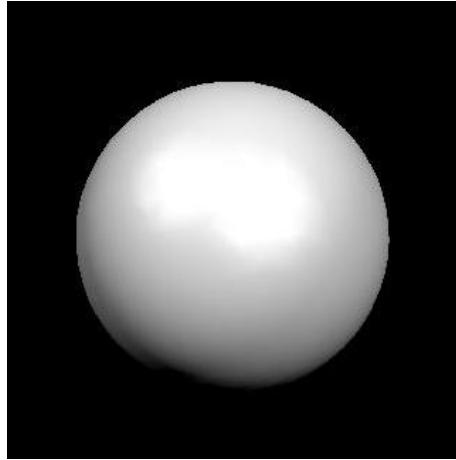


Is the contact density a good approximation to the effective density for heavy elements ?

# Relativistic atomic orbitals

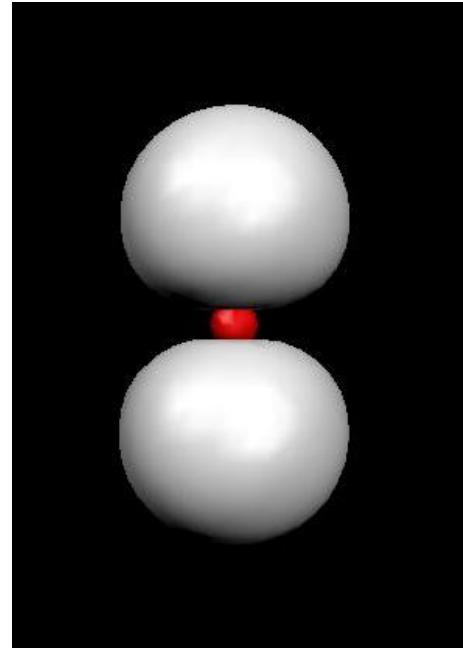
$$\psi^{\text{NR}}(r, \theta, \phi) = R(r)Y_{lm}(\theta, \phi) \Rightarrow \psi^{\text{R}}(r, \theta, \phi) = \begin{bmatrix} R^L(r)\chi_{\kappa, m_j}(\theta, \phi) \\ iR^S(r)\chi_{-\kappa, m_j}(\theta, \phi) \end{bmatrix}$$

	$s_{1/2}$	$p_{1/2}$	$p_{3/2}$	$d_{3/2}$	$d_{5/2}$
j	1/2	1/2	3/2	3/2	5/2
$\kappa$	-1	+1	-2	+2	-3



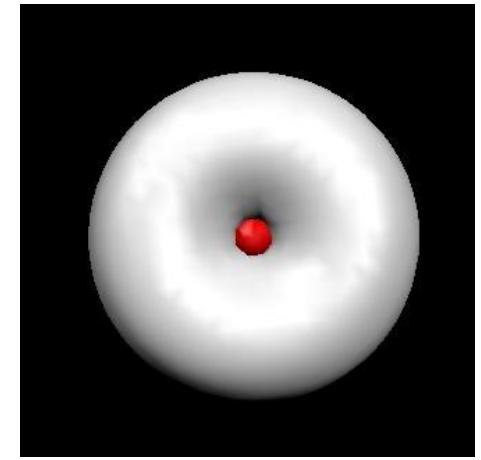
$2(s, p)_{1/2, 1/2}$

(z-axis anywhere)



$2(p, d)_{3/2, 1/2}$

(z-axis in plane)



$2(p, d)_{3/2, 3/2}$

(z-axis out of plane)

## Atomic contact density

$$\begin{aligned}\rho^{Hg}(0) &= \sum_i \left\langle \psi_i^{Hg} | \delta(\mathbf{r}) | \psi_i^{Hg} \right\rangle \\ &= \sum_i \left[ \left\langle R_i^L \left| \frac{\delta(r)}{4\pi r^2} \right| R_i^L \right\rangle_r \langle \chi_{\kappa,m} | \chi_{\kappa,m} \rangle_{(\theta,\phi)} + \right. \\ &\quad \left. + \left\langle R_i^S \left| \frac{\delta(r)}{4\pi r^2} \right| R_i^S \right\rangle_r \langle \chi_{-\kappa,m} | \chi_{-\kappa,m} \rangle_{(\theta,\phi)} \right] \\ &= \frac{1}{4\pi} \sum_i [R_i^L(0) R_i^L(0) + R_i^S(0) R_i^S(0)]\end{aligned}$$

Let us look at small  $r$  solutions of relativistic atomic orbitals

## Small r solutions

- We consider radial solutions at small  $r$ :

$$R^L = r^{\gamma-1} \left( p_0 + p_1 r + p_2 r^2 + \dots \right)$$

$$R^S = r^{\gamma-1} \left( q_0 + q_1 r + q_2 r^2 + \dots \right)$$

- Point nucleus:  $\gamma = +\sqrt{\kappa^2 - Z^2/c^2} < |\kappa|$  (singularity at the nucleus for  $|\kappa| = 1$ )
- Extended nuclei:  $\gamma = |\kappa|$ 
  - $\kappa < 0$ :  $q_0 = p_1 = 0$  (for  $s_{1/2}$ - orbitals ( $\kappa = -1$ )  $R^L(0) = p_0$  and  $R^S(0) = 0$ )
  - $\kappa > 0$ :  $p_0 = q_1 = 0$  (for  $p_{1/2}$ - orbitals ( $\kappa = 1$ )  $R^L(0) = 0$  and  $R^S(0) = q_0$ )

**In the relativistic case the contact density has contributions from the large components of  $s_{1/2}$  orbitals and the small components of  $p_{1/2}$  components.**

- Non-relativistic case:  $R(r) = r^l \left( a_0 + a_1 r + a_2 r^2 + \dots \right)$

( $a_1$  is zero for extended nuclei, thus removing the cusp)

## Contact vs. effective density

Orbital contributions for thallium atom (HF/TZ2s1p). All values are in atomic units  $a_0^{-3}$ .

	$\rho_e(0)$	$\bar{\rho}_e - \rho_e(0)$
$1s_{1/2}$	2112536.88	-215363.37
$2s_{1/2}$	322388.19	-33035.87
$3s_{1/2}$	74350.56	-7625.34
$4s_{1/2}$	18799.34	-1928.44
$5s_{1/2}$	3695.82	-379.14
$6s_{1/2}$	371.46	-38.11
$2p_{1/2}$	24695.53	-2449.09
$2p_{3/2}$	0	$2 \times 0.56$
$3p_{1/2}$	6395.44	-634.80
$3p_{3/2}$	0	$2 \times 0.16$
$4p_{1/2}$	1598.33	-158.67
$4p_{3/2}$	0	$2 \times 0.08$
$5p_{1/2}$	280.27	-27.82
$5p_{3/2}$	0	$2 \times 0.01$
$6p_{1/2}$	2.46	-0.24
$6p_{3/2}$	0	< 0.01
Total	2565114.28	-261639.27

## Contact vs. effective density gradient

Effective and contact electron density gradients, respectively, at the nuclei Tl, Pb, and Pt calculated at the four-component Dirac-Coulomb level using the TZ2s+1p basis. The derivatives (in  $\text{\AA}^{-4}$ ) are taken at the respective experimental geometries of TlI, PbTe, and PtSi.

Method	$\rho_0^{[1]X}$	$\bar{\rho}^{[1]X}$	$\Delta(\bar{\rho}^{[1]X} - \rho_0^{[1]X})$
TlI			
HF	212.15	190.29	-10.3%
DFT/PBE	114.53	102.67	-10.4%
DFT/SAOP	119.96	107.60	-10.3%
PbTe			
HF	240.18	215.00	-10.5%
DFT/PBE	155.74	139.29	-10.6%
PtSi			
HF	-1183.83	-1070.52	-9.6%
DFT/PBE	-656.39	-593.34	-9.6%

# Final results

method	Hamiltonian	$\rho^{[1]X}$	$V^X$
		TII	
experiment		120.6(38)	3.20(10)
DFT/SAOP	NR	21.09	0.57
CCSD(T)	<sup>4c</sup> DC**	142.26	3.81
DFT/SAOP (Cooke)	(ZORA)	21.1	0.61 <sup>d</sup>
PbTe			
experiment		148(11)	2.12(16)
DFT/SAOP	NR	19.64	0.28
CCSD(T)	<sup>4c</sup> DC**	163.19	2.36
DFT/SAOP (Cooke)	(ZORA)	21.1	0.33 <sup>d</sup>
PtSi			
experiment[C] <sup>e</sup>			-0.72(12)
DFT/SAOP	NR	-135.64	-1.14
CCSD(T)	<sup>4c</sup> DC**	-599.83	-5.05
DFT/SAOP (Cooke)	(ZORA)	-136.5	-1.10

All contact density results should be scaled down by 10%

## Conclusions and perspectives

- The effects of nuclear size can be observed in high-resolution rovibrational spectroscopy of diatomics of heavy atoms
- 4-component relativistic calibration studies show a curious coincidence of error in previous experimental and computational studies
- High accuracy work should employ the effective rather than the contact density
- Future work is to investigate the effect of nuclear size on vibrational constant
- <http://dirac.chem.vu.nl>

